## Biostimulation of Iron Reduction and Uranium Immobilization: Microbial and Mineralogical Controls



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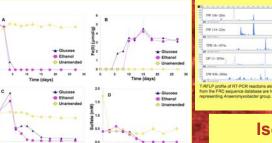
### **Abstract**

chanisms controlling the reduction and immobilization of U(VI) during timulation in shallow subsurface sediments cocontaminated with uranium nd nitrate. The focus is on the activity and community composition of microbial opulations (metal- and nitrate-reducing bacteria) and iron minerals that are kely to make strong contributions to the fate of U during *in situ* bioremediation. or integrated approach was applied to sediment cores and microcosms of site terials from Area 2 of the Field Research Center (FRC) at Oak Ridge, TN. ostantial differences were observed in the abundance and activity of microbia ips depending upon the electron donor (glucose or ethanol) used for tion. Viable counts revealed that Fe(III)- and nitrate-reducers are bundant (104 to 105 per g wet) in Area 2 sediments and counts were shown to e carbon substrate dependent. U(VI) and Fe(III) were reduced concurrently in e glucose but not the ethanol treatments. One to 2 orders of magnitude more III)-reducers were observed in ethanol- as compared to glucose-amended cultivatable Fe(III)-reducing bacteria in the ethanol treatments were numerically minated by Geobacter sp. while those cultured on glucose were dominated by entative organisms. Efforts are underway to associate in situ activity at the

ron minerals were characterized by Mössbauer spectroscopy over a wide inge in temperature (4 to 298 K) in order to fully determine the form and beciation of Fe. Spectra at room temperature (298 K) exhibited no sextet attern, thus excluding the presence of hematite, magnetite, and maghemite. A 7 K, the amount of Fe(II) doubled from 15 to 30 % in ethanol- and glucoseical extractions and counts of Fe(III)-reducing bacteria. Poorly ordered or ctivity. However, silicate bound Fe(III) clearly predominated over the Fe

Novel iron(III)- and sulfate-reducing organisms were isolated from the Ifotomaculum-related isolate utilizes Fe(III) as well as sulfate as an electron ceptor. The draft genome sequence of Geobacter strain FRC-32 has been impleted by the Joint Genome Institute and annotation is currently underway complete up the confine denoting institute and annotation is currently underway. Our results have the following implications for U bioremediation in the FRC subsurface: 1) the microbially-catalyzed mechanism of U(VI) reduction is electron donor dependent, 2) silicate bound Fe is an important oxidant that is transformed by indigenous microbial populations in the Area 2 subsurface, and cter sp. predominate over other Fe(III)-reducing bacteria during

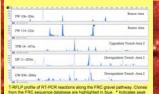
### **Microcosms**



Electron flow in microcosms of Area 2 sediments. Values are Fe(II), (c) U(VI), (d) sulfaté, (e)



### **Microbial Community Analysis**



GW 836. This is a DNA profile showing bacterial 16S



### **Conclusions**

### **Isolates**

- nuttie. The optimal pH was ~ 6 with no growth detected at pH 5

- The mechanism of U(VI) reduction is electron donor dependent, with substantial reduction occurring prior to
- Carbon utilization defined by fermentative metabolism and incomplete oxidation of ethanol
- Geobacter sp. numerically dominate cultivatable Fe(III)-
- metabolically active profiles of highly contaminated groundwaters collected near FRC source zone
- Highly impacted, upgradient groundwater samples had roughly half the T-RFLP peaks of downgradient samples suggesting lower diversity in contaminated groundwaters
- A database of > 1400 16S rRNA gene sequences retrieved from FRC materials is now available and > 400 have been examined with in silico digestion for T-RFLP
- Please see Denise Akob's poster for microbial community

- sediments for at least four groups of Fe(III)-reducing bacteria (Geobacter, Desulfotomaculum,
- Geobacter and Desulfotomaculum strains are physiologically distinct from close relatives. The maculum isolate is capable of growth using both
- The genome sequence of Geobacter strain FRC-32 is now available from JGI
- In collaboration with the Loeffler lab (Georgia Tech),

### Iron Mineralogy

- The following are estimated for goethite: a surface area of 42.5 m²/g, isomorphous Al substitution of 16.2%, and a
- Mössbauer supports wet chemical analysis to show

mean crystallite diameter (MCD) of 32.8 nm



# **Approach**

area 2 sediment (FB094) was combined ith Area 2 groundwater (FW209).

Treatments (3 replicates each): 20 mM Ethanol, 10mM Glucose, Unamended control

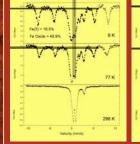
most probable number (MPN) dilution series was used to enumerate nitrate- and iron reducing bacteria present in the microcosms and FB094 sediment.

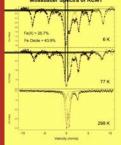
fter 3 months of growth the highest positive dilutions were sampled for NO<sub>3</sub>- and Fe(III)reduction, HPLC and molecular analysis (SSU rRNA).

itivation-Independent Community Analysis
NA was extracted from microcosm samples and MPN cultures.
oning and sequencing of PCR amplified SSU rRNA genes.

erprinting using terminal restriction fragment length polymorphism (TRFLP).
-SSU rRNA genes were PCR amplified with a fluorescently labeled 27F primer







**Iron Mineralogy** 

Sample	Phyllosilicate Peak Areas		Goethite Field & Peak Areas		Chemical Analysis*
	Aramo	Andro	Bur	A	Fe(II)
	*	%	Т	14	% of tot Fe
FB094	31.6	18.5	49.67	49.9	13.1
RCM7	37.8	12.9	49.72	49.3	12.5
RCM4	24.0	20.3	49.63	55.7	24.0
RCM1	30.3	25.7	49.83	43.9	31.2

MÖSSBAUER SPECTRA: obtained using a Web Research, Inc. spectrometer equipped with a Janis Model SHI-850-5 Closed Cycle Cryostat, operating at a sample temperature of 6 to 298 K.